

**Erratum: Semiempirical calculation of van der Waals coefficients and polarizabilities  
for the alkali-metal and alkaline-earth-metal atoms  
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**Erratum: Dispersion coefficients for H and He interactions with  
alkali-metal and alkaline-earth-metal atoms  
[Phys. Rev. A **68**, 062710 (2003)]**

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We have realized since publication that the higher- $\ell$  polarizabilities and dispersion coefficients are very sensitive to the representation of the ground-state wave function. The ground-state wave functions for Na and K were represented with a linear combination of Slater-type orbitals. One negative feature of a basis set expansion relates to the behavior at large distances from the nucleus. Unlike a grid based calculation, the correct asymptotics are not imposed and the larger- $r$  part of the wave function has a weak influence on the binding energy.

The ground-state wave functions for Na and K were both  $1.7 \times 10^{-6}$  hartree away from convergence. There were some small but significant changes in the polarizabilities and dispersion parameters when the existing basis sets were replaced by a large Laguerre basis set. The octupole polarizabilities decreased by 3–4 %, while  $C_{10}$  for dimers involving these atoms decreased by about 1–2 %.

TABLE I. Values of the polarizabilities and dispersion coefficients (in atomic units) for systems with Na or K.

| System | Property       | Present             | MBPT [1]           | Property       | Present             | MBPT [1]           |
|--------|----------------|---------------------|--------------------|----------------|---------------------|--------------------|
| Na     | $\alpha^{(2)}$ | 1879                | 1885               | $\alpha^{(3)}$ | $5.552 \times 10^4$ | $5.54 \times 10^4$ |
| K      | $\alpha^{(2)}$ | 5005                | 5000               | $\alpha^{(3)}$ | $1.777 \times 10^5$ | $1.77 \times 10^5$ |
| Na-H   | $C_8$          | 4012                |                    | $C_{10}$       | $2.916 \times 10^5$ |                    |
| Na-Li  | $C_8$          | $9.888 \times 10^4$ | $9.88 \times 10^4$ | $C_{10}$       | $9.187 \times 10^6$ | $9.16 \times 10^6$ |
| Na-Na  | $C_8$          | $1.159 \times 10^5$ | $1.16 \times 10^5$ | $C_{10}$       | $1.134 \times 10^7$ | $1.13 \times 10^7$ |
| Na-K   | $C_8$          | $2.240 \times 10^5$ | $2.24 \times 10^5$ | $C_{10}$       | $2.532 \times 10^7$ | $2.53 \times 10^7$ |
| Na-Rb  | $C_8$          | $2.649 \times 10^5$ | $2.66 \times 10^5$ | $C_{10}$       | $3.122 \times 10^7$ | $3.13 \times 10^7$ |
| K-H    | $C_8$          | 7970                |                    | $C_{10}$       | $7.345 \times 10^5$ |                    |
| K-Li   | $C_8$          | $1.955 \times 10^5$ | $1.95 \times 10^5$ | $C_{10}$       | $2.101 \times 10^7$ | $2.10 \times 10^7$ |
| K-K    | $C_8$          | $4.197 \times 10^5$ | $4.20 \times 10^5$ | $C_{10}$       | $5.374 \times 10^7$ | $5.37 \times 10^7$ |
| K-Rb   | $C_8$          | $4.904 \times 10^5$ | $4.93 \times 10^5$ | $C_{10}$       | $6.534 \times 10^7$ | $6.60 \times 10^7$ |
| Na-He  | $C_8$          | 1328                |                    | $C_{10}$       | $9.514 \times 10^4$ |                    |
| Na-Be  | $C_8$          | $3.349 \times 10^4$ |                    | $C_{10}$       | $2.651 \times 10^6$ |                    |
| Na-Mg  | $C_8$          | $6.736 \times 10^4$ |                    | $C_{10}$       | $5.743 \times 10^6$ |                    |
| Na-Ca  | $C_8$          | $1.633 \times 10^5$ |                    | $C_{10}$       | $1.561 \times 10^7$ |                    |
| Na-Sr  | $C_8$          | $2.172 \times 10^5$ |                    | $C_{10}$       | $2.183 \times 10^7$ |                    |
| K-He   | $C_8$          | 2623                |                    | $C_{10}$       | $2.398 \times 10^5$ |                    |
| K-Be   | $C_8$          | $6.514 \times 10^4$ |                    | $C_{10}$       | $6.382 \times 10^6$ |                    |
| K-Mg   | $C_8$          | $1.283 \times 10^5$ |                    | $C_{10}$       | $1.335 \times 10^7$ |                    |
| K-Ca   | $C_8$          | $3.024 \times 10^5$ |                    | $C_{10}$       | $3.462 \times 10^7$ |                    |
| K-Sr   | $C_8$          | $3.980 \times 10^5$ |                    | $C_{10}$       | $4.753 \times 10^7$ |                    |

The impact upon  $\alpha^{(2)}$  and  $C_8$  was an order of magnitude smaller and the changes were largely confined to the last digit of the revised data given in Table I. The impact upon  $\alpha^{(1)}$  and  $C_6$  was insignificant.

One feature of the revised calculation is that some small but annoying discrepancies with the many body perturbation theory (MBPT) results of Porsev and Derevianko [1] have been largely eradicated. The  $f^{(\ell)}$ -value distributions for Li and Rb were not affected by this problem since the  $2s$  and  $5s$  orbitals for these atoms were computed with a large Laguerre basis.

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[1] S. G. Porsev and A. Derevianko, J. Chem. Phys. **119**, 844 (2003).